

EFFECTIVE: SEPTEMBER 2004 CURRICULUM GUIDELINES

A.	Division:	Science and Technology		Effective Date:	September 2004							
В.	Department / Program Area:	Biology		Revision	X New Course							
				If Revision, Section(s) Revised:	C, H, M, R							
				Date of Previous Revision Date of Current Revision	<u> </u>							
C:	Biology 2421	D: Cel	l Biochemis	try	E: 3							
	Subject & Course No. Descrip			Title	Semester Credits							
F:	Calendar Description: An introduction to the biochemistry of a cell, including the structural and functional aspects of the micro and macro molecules. Protein structure, enzyme kinetics, and energy pathways will be considered. Some metabolic sequences in the cell will be examined in relation to control mechanisms.											
G:	/ Learning Settir	ls of Instructional Delivery		H: Course Prerequisites: Biology 2321 and Chemistry 320 or permission of the instructor.								
	Lecture/Tutoria		I:	Course Corequisites:								
	Number of Cont for each descript	act Hours: (per week / sem tor)	nester	Chemistry 420								
	Lecture: 3 hours/week Tutorial: 1 hour/week			J: Course for which this Course is a Prerequisite								
				None								
	Number of Wee	ks per Semester:										
	15 weeks			K: Maximum Class Size: 35								
L:	PLEASE INDIC											
	Non-Credi											
		redit Non-Transfer										
	X College Cr	redit Transfer:										
	SEE BC TRANS	SEE BC TRANSFER GUIDE FOR TRANSFER DETAILS (www.bccat.bc.ca)										

M: Course Objectives / Learning Outcomes

Upon completion of Biology 2421, the student will be able to:

- Describe the chemistry of water, acid-base properties, and buffers.
- Describe the chemistry of amino acids.
- Explain how protein sequence is determined, and describe the structure of peptides.
- Describe the structure of proteins, especially in terms of how this structure relates to function
- Describe what allosteric proteins are, and their importance.
- Describe the structure, function, and behaviour of hemoglobin and myoglobin.
- Describe enzyme kinetics
- Explain basic bioenergetic principles as they relate to catabolism in the cell free energy, coupled reactions, nucleotides.
- Describe the chemistry of carbohydrates structure and function
- Explain in detail the process of cellular respiration glycolysis, Krebs cycle, electron transport and ATP synthesis.
- Describe anabolism in the cell in terms of gluconeogenesis.
- Describe the biosynthesis of macromolecules (specifically polysaccharides) in terms of glycogen synthesis, and describe the degradation of macromolecules in terms of glycogenolysis.
- Describe metabolic control in the cell and energy charge.
- Describe regulation in the cell in terms of hormone action.
- Provide brief descriptions of alternative oxidative pathways i.e. lipid and fatty acid oxidation, amino acid oxidation, the phosphogluconate pathway.
- Provide a brief overview of human metabolism in terms of interrelationships between the catabolic and anabolic pathways discussed during the course of the semester.

N: Course Content:

The major topics in the course include the following:

- 1. An Introduction What is Biochemistry?
- 2. Proteins:
 - Water and Acid-Base concepts
 - Amino acids, peptides, and proteins
 - The Henderson-Hasselbalch Equation
 - PH, pK, and pI.
 - Electrophoresis
 - Peptide sequencing
 - Protein structure
 - Titration curves of amino acids and peptides

Course Content: (cont.)

3. Globular proteins

- Myoglobin (Mb) structure, function, and behaviour
- Hemoglobin (Hb) structure, function, and behaviour
- Major differences between myoglobin and hemoglobin
- Adult hemoglobin versus fetal hemoglobin
- The effect of certain metabolites (i.e. H+ ions, CO, and BPG) on hemoglobin
- Sickle cell anemia and its effect on hemoglobin structure and function

4. Enzyme Kinetics

- Enzymes as biological catalysts
- Reaction rates
- The specificity of enzymes for their substrates
- Specific catalytic groups and their contribution to catalysis
- Substrate concentrations
- The Michaelis-Menten Equation
- Lineweaver-Burk plots
- The meaning of Vmax and Km as they relate to enzymes
- Reversible and irreversible inhibition
- The affect of pH on enzyme activity
- Allosteric enzymes, and how their kinetics differ from those of non-allosteric enzymes

5. Bioenergetics

- The Laws of Thermodynamics a short review
- Entropy
- Standard and Actual Free-Energy Change
- The Equilibrium Constant
- Coupled reactions
- Phosphate group transfers and ATP

6. Catabolism

- Glycolysis
- The Tricarboxylic Acid Cycle or Krebs Cycle
- The Electron Transport System
- The Glycerol-Phosphate and Malate-Aspartate Shuttle Mechanisms
- Gluconeogenesis
- Glycogen metabolism Glycogen synthesis and Glycogenolysis
- Other alternative oxidative pathways
- The effect of hormones on metabolism
- Integration of metabolism

O: Methods of Instruction

This course involves three hours a week of classroom instruction and one hour a week of tutorials in which selected problems from the textbook are solved.

P: Textbooks and Materials to be Purchased by Students

Nelson and Cox, 2000. Lehninger – Principles of Biochemistry (3rd edition). New York: Worth Publishers.

Q:	Means	of Assessmo	ent							<u> </u>
	TYPE OF EVALUATION				POINTS					
		Class tests			10-25					
	Two term examinations			30-50						
	One final examination TOTAL			30-40 100						
		Grades:	A+ 95-100, C+ 65-69,						75-79, 0-49	В- 70-74,
R:	Prior Learning Assessment and Recognition: specify whether course is open for PLAR									AR
	There is no provision of PLAR, other than that normally done by examining transcripts and comparing course outlines of Biochemistry courses taken within the last five years elsewhere to the Douglas College Biology 2421 course content.									
Cours	se Designer	r(s)			Edu	cation	Council	Cur	riculum C	ommittee Representative
Dean / Director				Reg	Registrar					

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