

EFFECTIVE: SEPTEMBER 2003 CURRICULUM GUIDELINES

| А. | Division: Science and Technology | Ef | fective Date: Septen | nber 200 | 3 | |
|----|--|----------|--|---------------|----------------------|------|
| B. | Department / Program Area: Biology | If | vision Revision, Section(s) vised: A,B,F,G,M,N | X ,O,P,Q,F | New Course | |
| | | Da | te of Previous Revis | ion: Sept | ember 1984 | |
| | | | te of Current Revisio | | | |
| C | | | | JII. Ividy | | |
| C: | Biology 421 D: Cell Biochen | nistry | | | E: 3 | |
| | Subject & Course No. Descript | tive Ti | tle | Ser | nester Credits | |
| F: | Calendar Description: | | | | | |
| | An introduction to the biochemistry of a cell, include macro molecules. Protein structure, enzyme kinetics metabolic sequences in the cell will be examined in | s, and e | energy pathways will | l be cons | | nd |
| G: | Allocation of Contact Hours to Type of Instruction / Learning Settings | H: | Course Prerequisite Biology 321 and C | | y 320, or permission | n of |
| | Primary Methods of Instructional Delivery and/or Learning Settings: | | the instructor. | 5 | | |
| | Lecture/Tutorial/Lab | I: | Course Corequisite | es: | | |
| | | | Chemistry 420 | | | |
| | Number of Contact Hours: (per week / semester | | | | | |
| | for each descriptor) | J: | Course for which t | his Cour | se is a Prerequisite | |
| | Lecture: 3 hours/week | | None | | | |
| | Tutorial: 1 hour/week | | | | | |
| | | | Maximum Class S | ize: | | |
| | Number of Weeks per Semester: 14 | | 35 | | | |
| L: | PLEASE INDICATE: | <u> </u> | | | | |
| | Non-Credit | | | | | |
| | College Credit Non-Transfer | | | | | |
| | X College Credit Transfer: | | | | | |
| | SEE BC TRANSFER GUIDE FOR TRANSFER DE | TAII | S (www.bccat.bc.ca) | | | |

| M: | Course Objectives / Learning Outcomes |
|-----------|--|
| | Upon completion of Biology 421, the student will be able to: Describe the chemistry of water, acid-base properties, and buffers. Describe the chemistry of amino acids. Explain how protein sequence is determined, and describe the structure of peptides. Describe the structure of proteins, especially in terms of how this structure relates to function. Describe what allosteric proteins are, and their importance. Describe the structure, function, and behaviour of hemoglobin and myoglobin. Describe enzyme kinetics Explain basic bioenergetic principles as they relate to catabolism in the cell – free energy, coupled reactions, nucleotides. Describe the chemistry of carbohydrates – structure and function. Explain in detail the process of cellular respiration – glycolysis, Krebs cycle, electron transport and ATP synthesis. Describe the biosynthesis of macromolecules (specifically polysaccharides) in terms of glycogen synthesis, and describe the degradation of macromolecules in terms of glycogenolysis. Describe metabolic control in the cell in terms of hormone action. Provide brief descriptions of alternative oxidative pathways – i.e. lipid and fatty acid oxidation, amino acid oxidation, the phosphogluconate pathway. Provide a brief overview of human metabolism in terms of interrelationships between the catabolic and anabolic pathways discussed during the course of the semester. |
| N: | Course Content: |
| | The major topics in the course include the following: |
| | An Introduction – What is Biochemistry? |
| | Proteins: |
| | Water and Acid-Base concepts |
| | Amino acids, peptides, and proteins |
| | The Henderson-Hasselbalch Equation PH, pK, and pI. |
| | Electrophoresis Peptide sequencing |
| | Protein structure |
| | Titration curves of amino acids and peptides |
| | Globular proteins |
| | Myoglobin (Mb) – structure, function, and behaviour |
| | Hemoglobin (Hb) – structure, function, and behaviour Major differences between myoglobin and hemoglobin |
| | Adult hemoglobin versus fetal hemoglobin |
| | The effect of certain metabolites (i.e. H+ ions, CO, and BPG) on hemoglobin Sickle cell anemia and its effect on hemoglobin structure and function |
| | Enzyme Kinetics |
| | Enzymes as biological catalysts |
| | Reaction rates The specificity of enzymes for their substrates |
| | Specific catalytic groups and their contribution to catalysis |
| | Substrate concentrations The Michaelis-Menten Equation |
| | Lineweaver-Burk plots |

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|---------------|--|
| | The meaning of Vmax and Km as they relate to enzymes |
| | Reversible and irreversible inhibition |
| | The affect of pH on enzyme activity Allosteric enzymes, and how their kinetics differ from those of non-allosteric enzymes |
| | Anosteric enzymes, and now men kinetics differ from those of non-anosteric enzymes |
| | Bioenergetics |
| | The Laws of Thermodynamics – a short review |
| | Entropy |
| | Standard and Actual Free-Energy Change |
| | The Equilibrium Constant |
| | Coupled reactions |
| | Phosphate group transfers and ATP |
| | Catabolism |
| | Glycolysis |
| | The Tricarboxylic Acid Cycle or Krebs Cycle |
| | The electron Transport System |
| | The Glycerol-Phosphate and Malate-Aspartate Shuttle Mechanisms Gluconeogenesis |
| | Glycogen metabolism – Glycogen synthesis and Glycogenolysis |
| | Other alternative oxidative pathways |
| | The effects of hormones on metabolism |
| | Integration of metabolism |
| | |
|): | Methods of Instruction |
| | Methods of Instruction This course involves three hours a week of classroom instruction and one hour a week of tutorials in which selected problems from the textbook are solved. Textbooks and Materials to be Purchased by Student |
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Dean / Director

Registrar

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